BASIC N-P-K LIQUID ISSUES

MICHAEL ORR SPEIALTY PROCESS CONSULTING Fluid Fertilizer Foundation DECEMBER 2012





Materials

Cold vs. hot mixing

Potential issues with liquid blends

Example

Base Liquid Materials

Water

Urea Ammonium-Nitrate (UAN)

Ammonium Polyphosphate (10-34-0)

■ Solution grade muriate of potash (0-0-62)

Additional Liquids

- Wet process phosphoric acid
 Potassium thiosulfate (KTS)
 Aqua ammonia
- Urea
- □ Ammonium Thiosulfate (12-0-0-26(S))

Factors Affecting Blend Quality

Desired salting out temperature
 Storage temperature desired
 Impurities in materials or storage tanks
 Polyphosphate content

Factors Affecting Cold Mixing

Temperature of raw materials.

- Colder materials take longer to mix.
- Potash cools solution on addition.
- Can use hot water to speed mixing times.
- Salting out temperature of final product.
 - Higher salting out temperature requires longer mix times.



Hot Mixing

Heat supplied by chemical reaction: Strong acid, phosphoric acid (merchant grade) Strong base, aqua ammonia or anhydrous Must use 10-34-0 to supply polyphosphate Without potash, • a blend requires 25% of P_2O_5 from 10-34-0 With potash □ must have 35 - 45% of P₂O₅ from 10-34-0

Advantages of Hot Mixing

No external heat source required.
Faster mixes.

- Cheaper mixes made using merchant grade phosphoric acid and aqua ammonia than just 10-34-0.
- Can use dry urea as N source since heat will help dissolve urea.



Calculating Liquid Formulas

Cold mixing steps:

- 1. Determine application rate.
- 2. Calculate N:P:K ratio.
- 3. Convert fractional ratio to integer ratio.
- 4. Refer to table to determine grade.
- 5. Calculate formula.
- 6. Calculate application rate.

Cold Mix Example Step 1: Application rate

Desired application rate:

■ N 60 lbs./acre

 $\square P_2O_5 \quad 40 \text{ lbs./acre}$

 \bullet K₂O 40 lbs./acre

Using UAN-32, 10-34-0, 0-0-62 & water

Step 2: N: P₂O₅: K₂O Ratio

Divide smallest rate into other rates \sim N: 60/40 = 1.5 $\square P_2O_5: 40/40 = 1.0$ • $K_2O: 40/40 = 1.0$ Ratio is 1.5:1:1

Step 3: Convert to Integer

In this case, multiply all numbers by 2 • $1.5 \times 2 = 3$ • $1.0 \times 2 = 2$ • $1.0 \times 2 = 2$ The resulting ratio is 3:2:2

Step 4: Determine Grade Refer to Cold Mix Table ■ For 3:2:2 the grade that will have a salt out temperature of 32°F is: ■ 8.7% N \bullet 5.8% P₂O₅ ■ 5.8% K₂O

Step 5a: Calculate Lbs./Ton

Multiply all percentages by 20

• N: $8.7 \times 20 = 174 \text{ lbs./ton}$

 \square P₂O₅: 5.8 x 20 = 116 lbs./ton

• K_2O : 5.8 x 20 = 116 lbs./ton

Step 5b: Weight of Material/ton

\Box (116 lbs./ton)/0.34 = 341 lbs. of 10-34-0/ton **UAN-32:** ■ N required from UAN-32 after 10-34-0 $= 174 - (341 \times 0.10) = 139.9$ lbs. ■ Weight/ton = 139.9/0.32 = 437 lbs./ton 0-0-62: \square 116/0.62 = 187 lbs./ton

Step 5c: Weight of Water Subtract total weight of other materials from 2,000 lbs.

2,000 - (341+437+187) = 1,035 lbs. Water/ton

Step 5d: Summary Formula

Step 6: Calc. Application Rate Determine lbs. blend/acre 60/0.087 = 690 lbs. blend/acre Determine gallons/acre Assume most blends weigh 10 lbs./gallon (690 lbs./acre)/(10 lbs./gallon) = 69 gpa

Hot Mix Example Step 1: Application rate

Desired application rate:

- □ N 25 lbs./acre
- \square P₂O₅ 75 lbs./acre
- K_2O 25 lbs./acre

Using Phos acid, aqua, UAN-32, 10-34-0, 0-0-62 & water

Step 2: N: P₂O₅: K₂O Ratio

Divide smallest rate into other rates

N: 25/25 = 1P₂O₅: 75/25 = 3K₂O: 25/25 = 1Ratio is 1:3:1

Step 4: Determine Grade

Refer to Hot Mix Table
For 1:3:1 the grade that will have a salt out temperature of 32°F is:
7.4 % N
22.2 % P₂O₅

• $7.4 \% \text{ K}_2\text{O}$

Step 5: Calculate Lbs./Ton

Multiply all percentages by 20

• N: $7.4 \times 20 = 148 \text{ lbs./ton}$

 \square P₂O₅: 22.2 x 20 = 444 lbs./ton

• K_2O : 7.4 x 20 = 148 lbs./ton

Step 6: Weight of 10-34-0

Need to have 45% of P_2O_5 from 10-34-0 Total $P_2O_5 \ge 0.45 =$ weight of polyphosphate P_2O_5 $444 \ge 0.45 = 200$ lbs. 10-34-0 is 70% polyphosphate so: 200/0.70 = 286 lbs. of P₂O₅ from 10-34-0 Now calculate Lbs./ton of 10-34-0 286/34 = 841 lbs./ton

Step 7: Lbs. P₂O₅ from phos acid

Need a total of 444 lbs. P_2O_5 10-34-0 supplies 286 lbs. P_2O_5 444 – 286 = 158 lbs. P_2O_5 Lbs./ton 158/0.54 (54% acid) = 293 lbs. Phos acid

Step 8: Weight of aqua

Need to ammoniate acid to N: P_2O_5 ratio of 3:10. Phos acid added at 158 lbs./ton $158 \times 0.3 = 47.4$ lbs. N Lbs. Aqua/ton 47.4/0.2 = 237 lbs aqua/ton

Step 9: Weight of UAN per ton

Total N – (N from aqua + N from 10-34-0) 148 – [47.4 +(841 x 0.1)] = 16.5 lbs. N Lbs. UAN/ton 16.5/0.32 = 52 lbs. UAN/ton

Step 10: Weight of KCl per ton

Require 148 lbs K_2O per ton 148 lbs./0.62 = 238 lbs. KCl/ton

Step 11: Weight of water

2000 lbs. – (weight of other materials) 2,000 – (841 + 293 + 237 + 52 + 238) = 339 lbs. Water/ton

Summary Formula

For 7.4 – 22.2 – 7.4 grade liquid: 339 lbs. Water Phos acid 293 lbs. 237 lbs. Aqua UAN 52 lbs. 10-34-0 841 lbs. 0-0-62 238 lbs. 2,000 lbs. Total =

GENERAL BLENDING PARAMETERS

- MAXIMUM N-P-K IN SOLUTION 35%
- MICROS MAXIMUM IN SOLUTION 6%
- POLY/ORTHO PHOSPHATE BLENDS?
- LOW SALT BLENDS 35% DEPENDING ON RATIO

UAN/10-34-0 BLEND

NITRATE/ NO_X

NO WITH ACIDS SIGNS REDDISH/BROWN GAS NERVE GAS/DEADLY NITRATE NOT HIGHLY SOLUBLE WITH POTASSIUM WILL COMPETE WITH OTHER ANIONS IN SOLUTIONS KEEP IN MIND WITH UAN SOLUTION MIXING

Secondary and Micro Nutrients

Good Quality Micro Products Improve Blending Pay Attention to Order Of Addition, pH, etc Be Careful Of Point Introduction Precipitation Make Sure Adequate Mixing Time Prior To Loading ? Truck Mixing Of Blends

Receiving and Unloading Materials Into the PLANT

 HAVE ALL PERTINENT SHIPPING INFORMATION, BOLs, DOT requirement, etc.
 REVIEW AND UNDERSTAND MSDS AND RMP/PSM REQUIREMENTS
 HAZARDOUS MATERIALS RECEIVING

Filling Liquid Storage Tanks

- Make Sure Inbound Transport Is Connected to the Correct Storage Tank
- Open All Appropriate Valves and Close Others
- Contain All Leaks
- Check Tank Inventory Prior To Transfer
- Wear Proper PPE
 - Close All Appropriate Valves Upon Completion
 Of Transfer
- Complete All Documentation and Record Ending
 Inventory

COMMON MICRONUTRIENTS

Zinc Manganese Copper ■ Iron Magnesium Boron

Atomic Number: 26 Atomic Mass: 55.85

(Mg)

B Boron Atomic Number: 5 Atomic Mass: 10. 81

Oxides, Sulfates, and Chelates

- Zinc Oxide, Zn Sulfate, Zinc Citrate, and Zinc EDTA
- Manganese Dioxide, Manganese Sulfate, and Manganese EDTA
- Copper Sulfate, Copper Citrate, and Copper EDTA
- Ferrous Sulfate, and Iron EDTA
- Manganese EDTA
- Boric Acid

Conclusion

Wide range of blends possible!
 Pull away from the pack!
 Use soil and tissue tests to optimize your liquid program to the crop.

QUESTIONS

